

Unique characteristics of the novel carbonaceous film with high electrical conductivity and ultra high hardness for semiconductor test probes



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Outline

1. What is the carbonaceous film?

- Introduction

2. Can the film actually be used productively in the industry?

- Industrial issue

3. Why does this film have these characteristics?

- Discussion

4. Are there any care points of practical use?

- Care points of practical use

5. Conclusion

Introduction

1. WHAT IS THE CARBONACEOUS FILM?

The carbonaceous film has 3 outstanding properties.

Higher hardness: Hv 4000

> 20 times higher than Au
> 2 times higher than conventional film



Longer life time!

Lower contamination of
electrode dust

Self-cleaning surface



**Higher stability of
conductivity!**

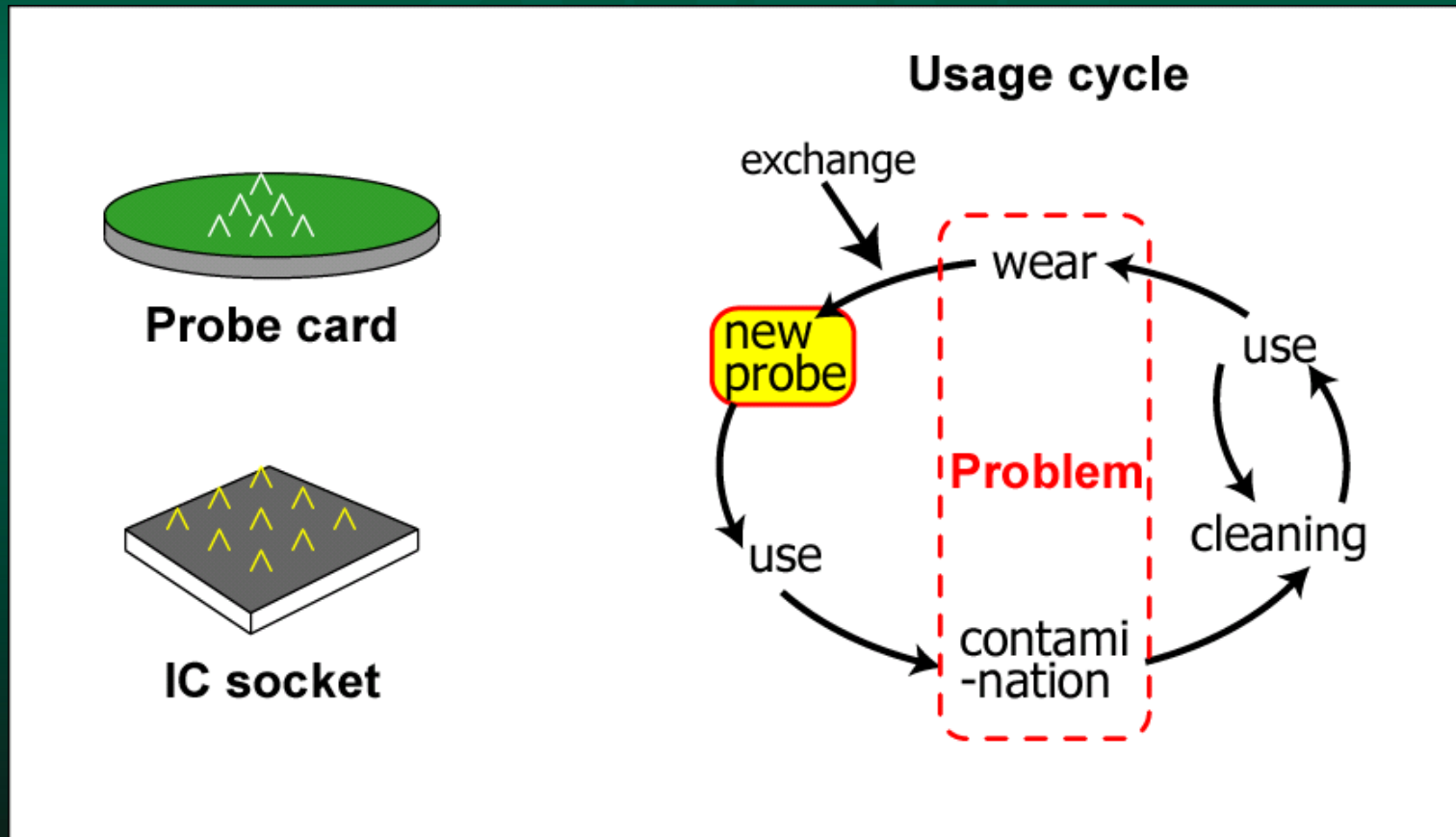
Steel = 1×10^{-4} ohm cm
Conductive DLC = 5×10^{-3} ohm cm

**High electrical conductivity
= 5×10^{-3} ohm cm**



**As good as metal
conductivity!**

Industry needs a probe surface with wear-resistance, and electrical stability.



Problem -> Solution

- Wear -> High hardness (wear-resistance)
- Contamination (decrease of electrical stability) -> Self-cleaning surface

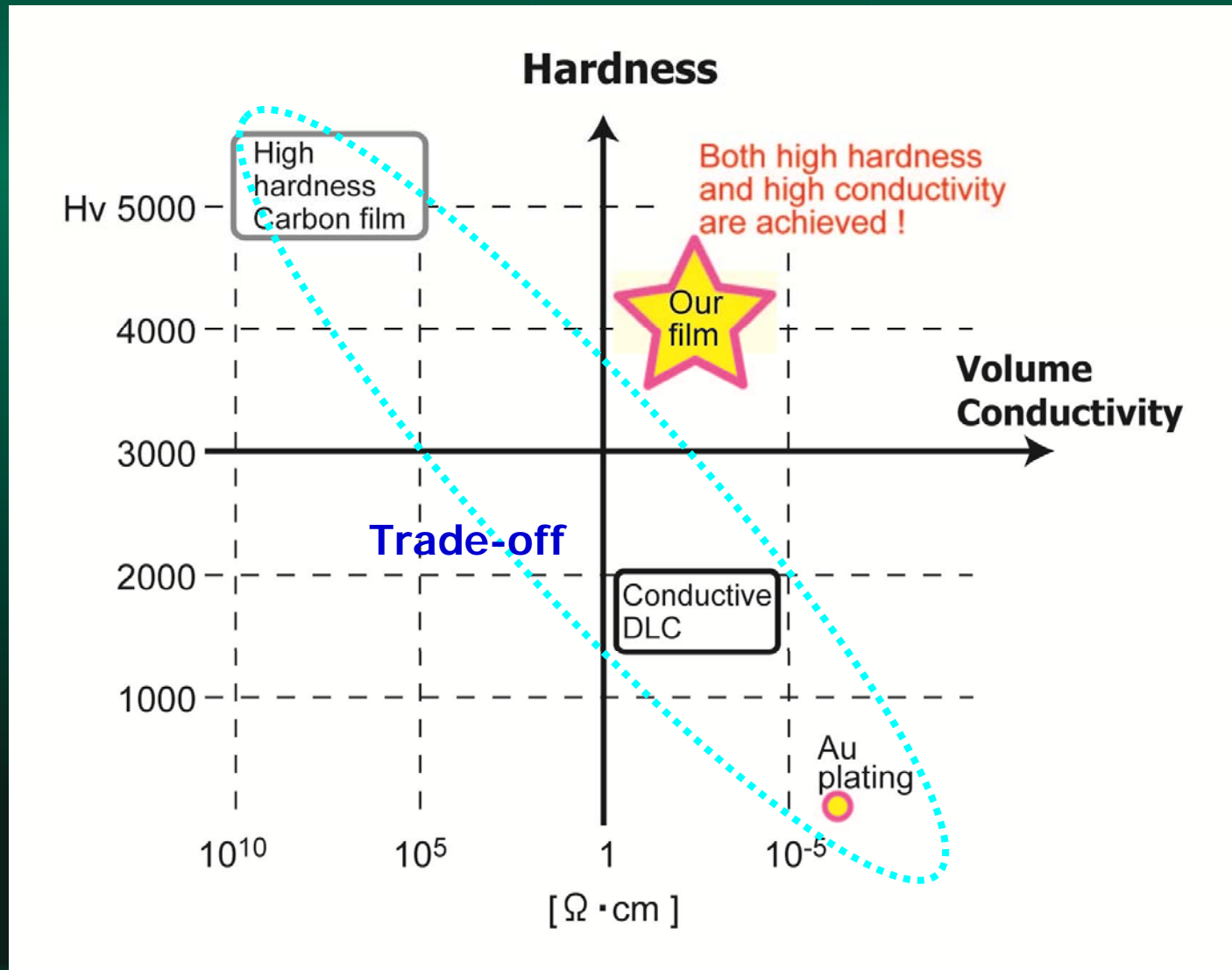
The film has the potential as future surface treatment for probes.

Surface Treatment	Hardness	Self-cleaning surface	Electrical Conductivity
Au Plating	X ($< \text{Hv } 250$)	X	○
Palladium Alloy Plating	△ ($\text{Hv } 350 \sim 400$)	X	○
Rhodium Plating	△ ($\text{Hv } 800 \sim 900$)	X	○
Conductive DLC (Diamond-Like Carbon)	△ ($\text{Hv } 1000 \sim 2000$)	△	△
Our film	○ ($\text{Hv } 4000$)	○	△

[Key]

- = very effective
- △ = somewhat effective
- X = ineffective

The film has an advantage compared with prior treatment.



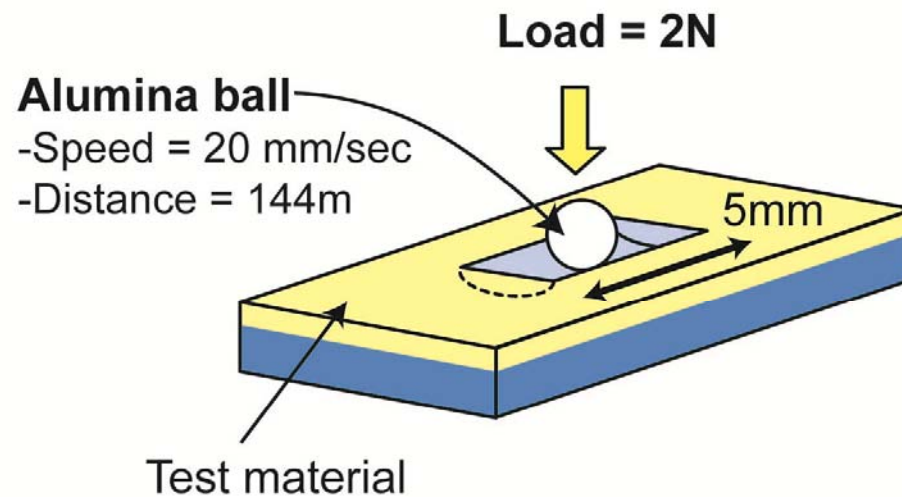
Industrial issue

2. CAN THE FILM ACTUALLY BE USED PRODUCTIVELY IN THE INDUSTRY?

Yes, it can. Why?

- **Because, the film has 3 points for customer appeal.**
 - (a) Higher wear-resistance
 - (b) High electrical & more stable conductivity
 - (c) Self cleaning surface

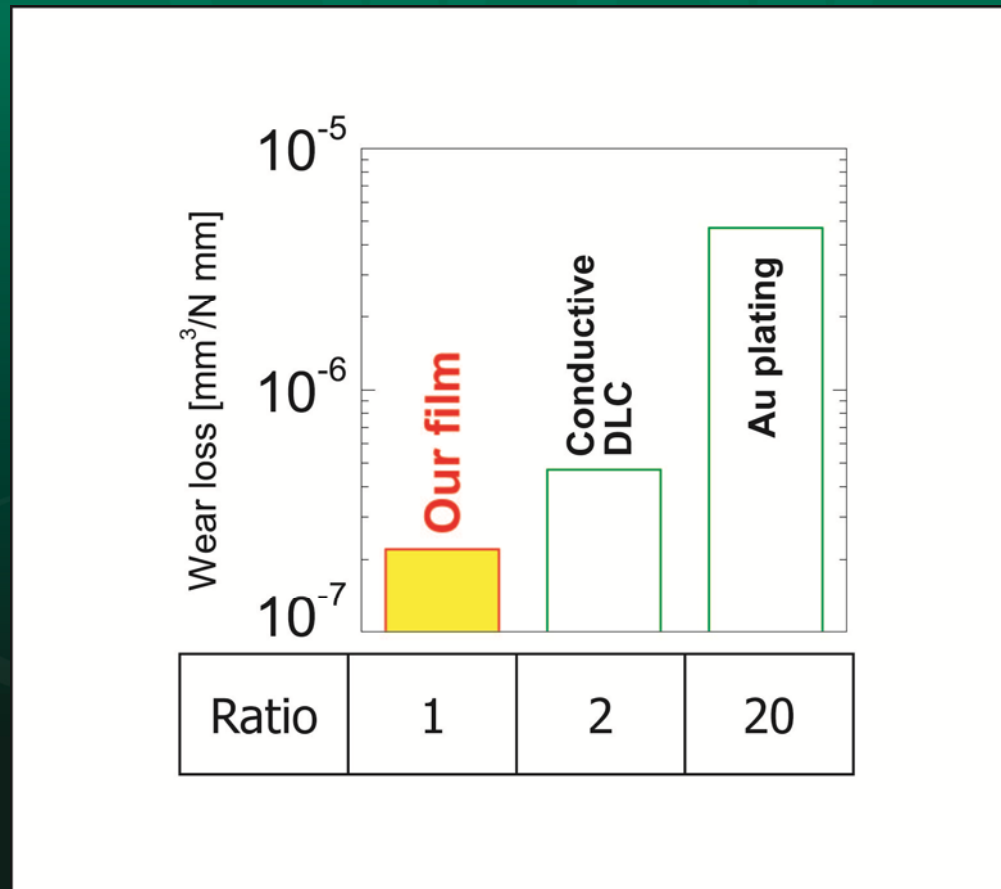
(a-1-1) The method of wear-loss test



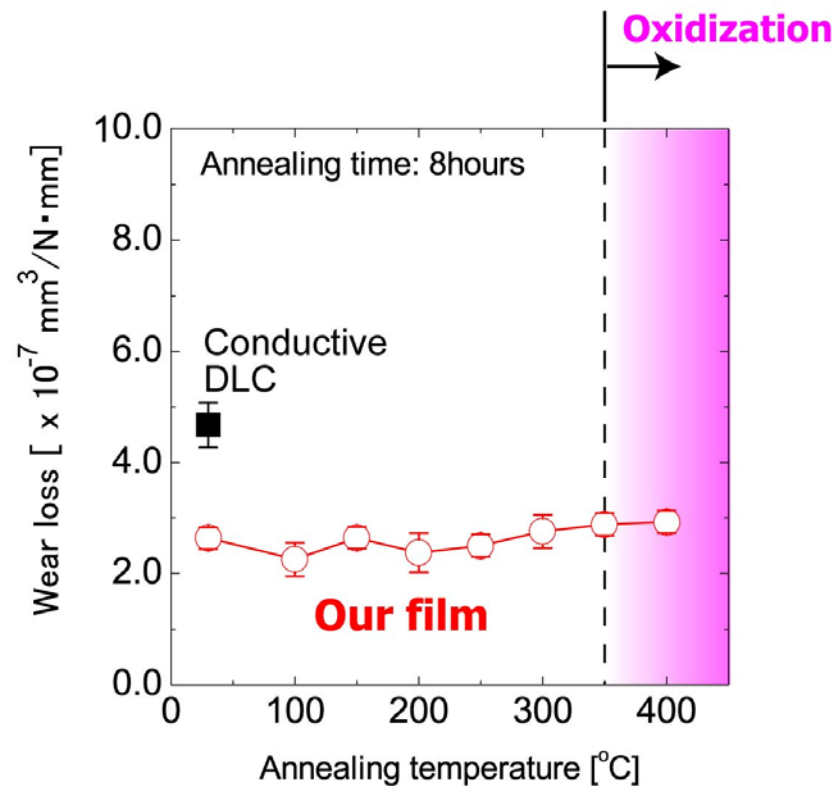
Schematic diagram of wear-loss test

(a-1-2) The higher wear-resistance is obtained compared with alternative materials.

- Our film has the minimum wear-loss after friction test.

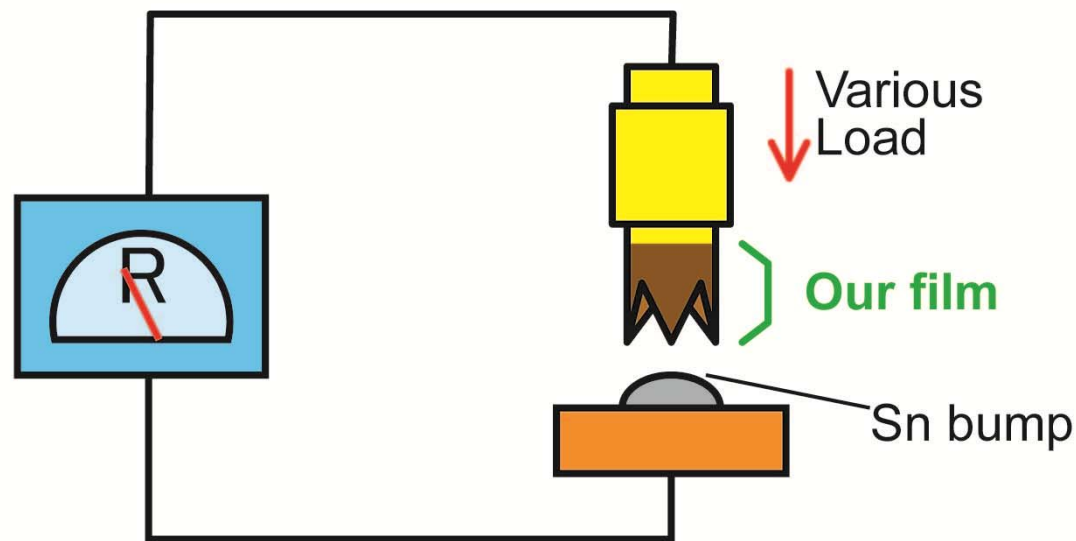


(a-2) The higher wear-resistance is maintained up to 350 °C.



Annealing temperature dependence on wear loss for 8 hours.

(b-1-1) Method of contact resistivity measurement of the film.



Schematic diagram of contact resistivity measurement

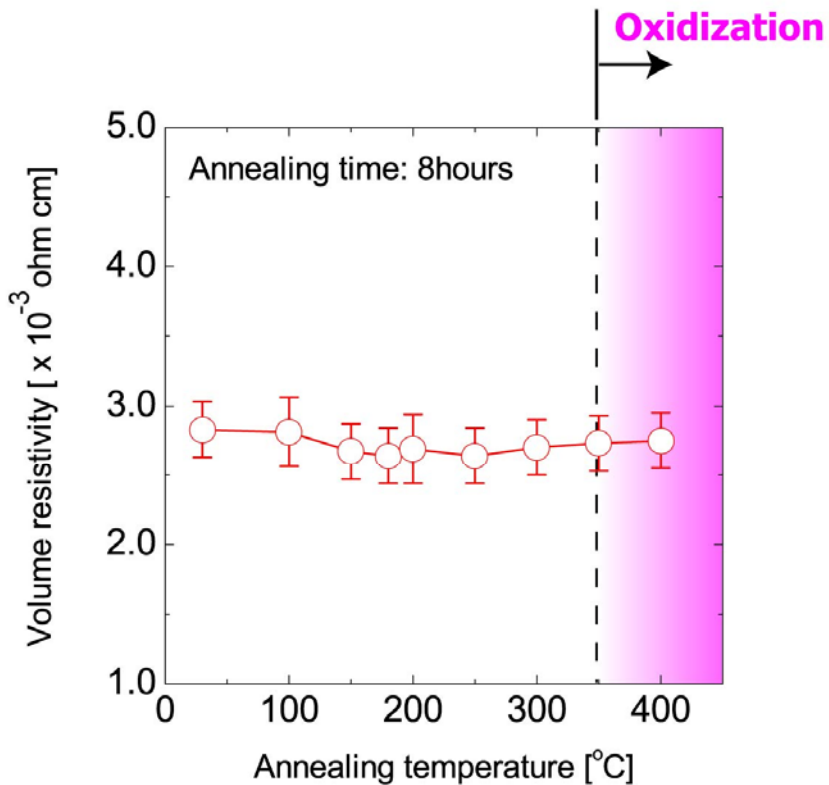
(b-1-2)

High electrical conductivity is obtained at low load.

- Contact resistivity below 200m ohm is obtained with 10gf Load.

	Our film@2011	Our film@2012	Au plating
Contact resistivity with 10gf	> 2 ohm	0.12 ohm	0.10 ohm
Load vs. Contact resistivity			

(b-2) High electrical conductivity is maintained up to 350 °C.



Annealing temperature dependence on volume resistivity for 8 hours

“From experimental results of touchdowns with Sn bump”

1) Conductive DLC

-> Increase in resistivity above 100 °C, because of the oxidization of the metal component in the film.

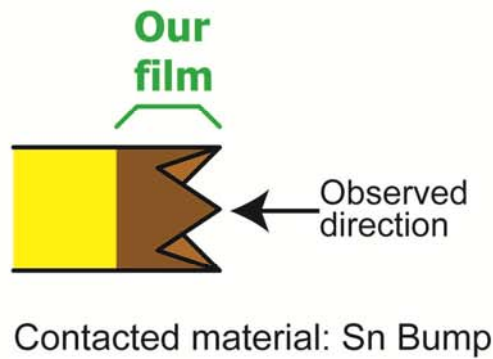
2) Au, Palladium plating

-> Increase in resistivity above 100 °C, because of the contamination on the probe surface.

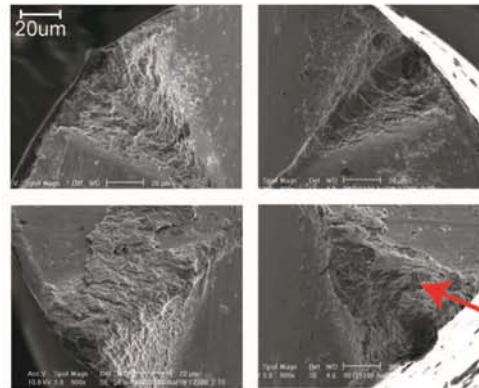
3) Our film

-> **Stable resistivity above 100 °C.**

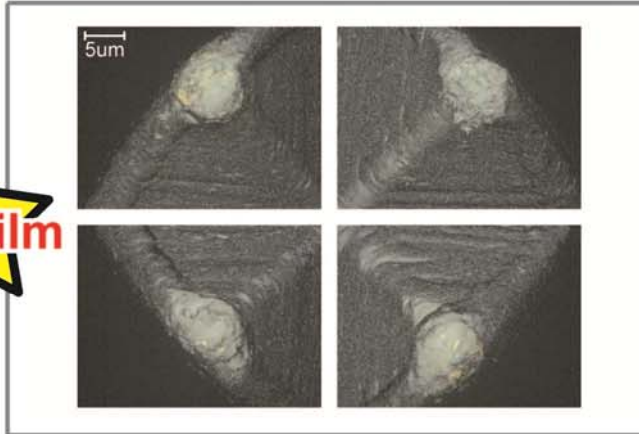
(c) Self-cleaning surface is maintained, after 30k touchdowns.



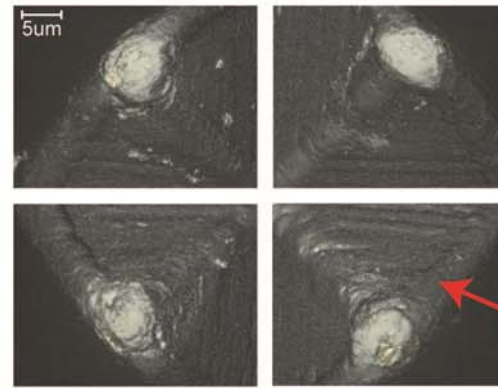
Au plating



Our film



Before



After 30k touchdowns
 -Contact temp.: 125°C
 -No-cleaning with a pad

Discussion

3. WHY DOES THIS FILM HAVE THESE CHARACTERISTICS?

- (a) Higher wear-resistance
- (b) High electrical & more stable conductivity
- (c) Self cleaning surface

(a) Higher wear-resistance is obtained with structural stability of the carbonaceous film.

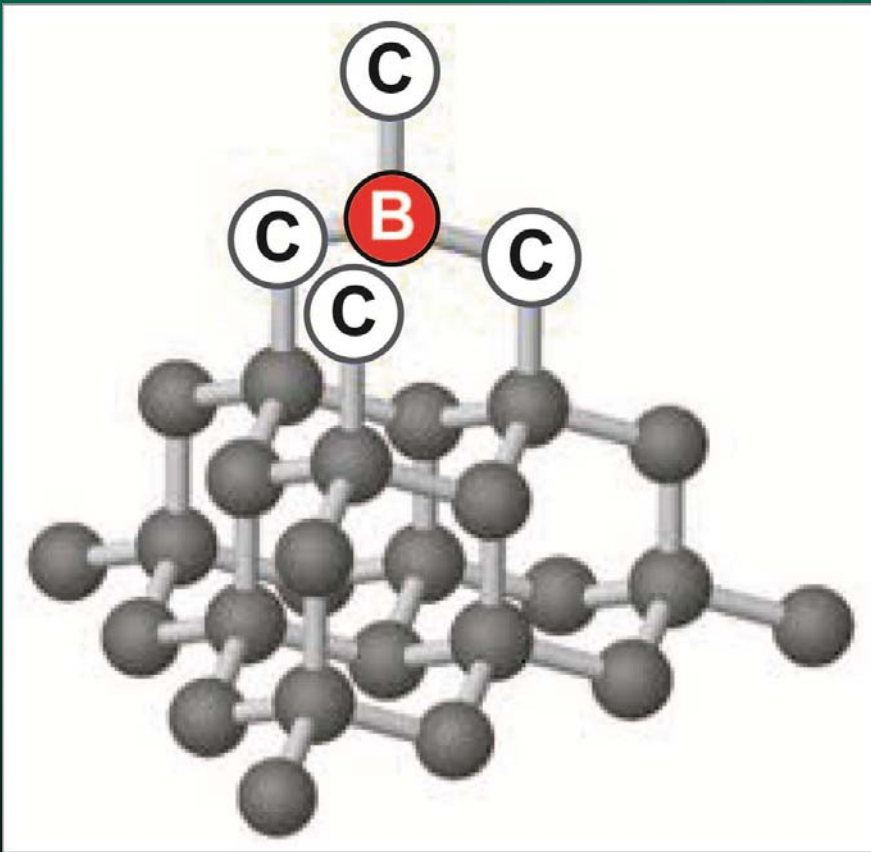
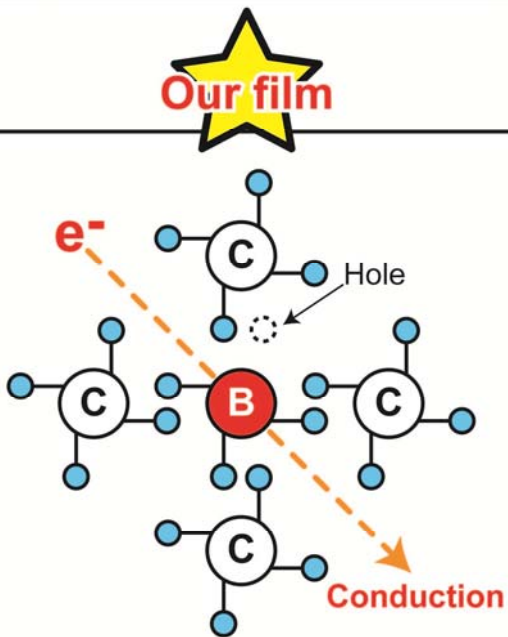
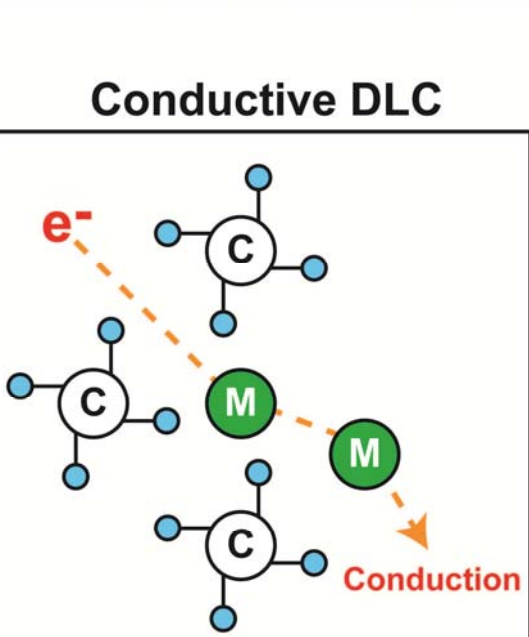


Image of the film structure

- The structural stability is caused by 3 reasons.
 - (1) Base material is carbon.
 - (2) The structure consists of diamond-like bonding.
 - (3) Boron, which is doped for the conduction, is less than 1%.

(b) High electrical & more stable conductivity is given by low content of Boron.

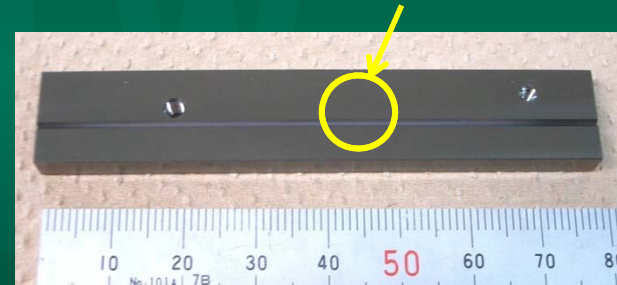
	Our film	Conductive DLC
Conduction model		
Principle of conduction	p-type conduction of covalent bond	Metal conduction
Contents of the element	B < 1 atomic%	Metal < 20~40 atomic%
Stability against the oxidization	Stable up to 350°C	Unstable above 100°C

(c) Self-cleaning surface is obtained by lower surface free energy of the film.



Example of surface free energy

Lower contamination of Sn



Picture: Parts feeder for MLCC deposited with the carbonaceous film. (Multi-Layer Ceramic Condenser)

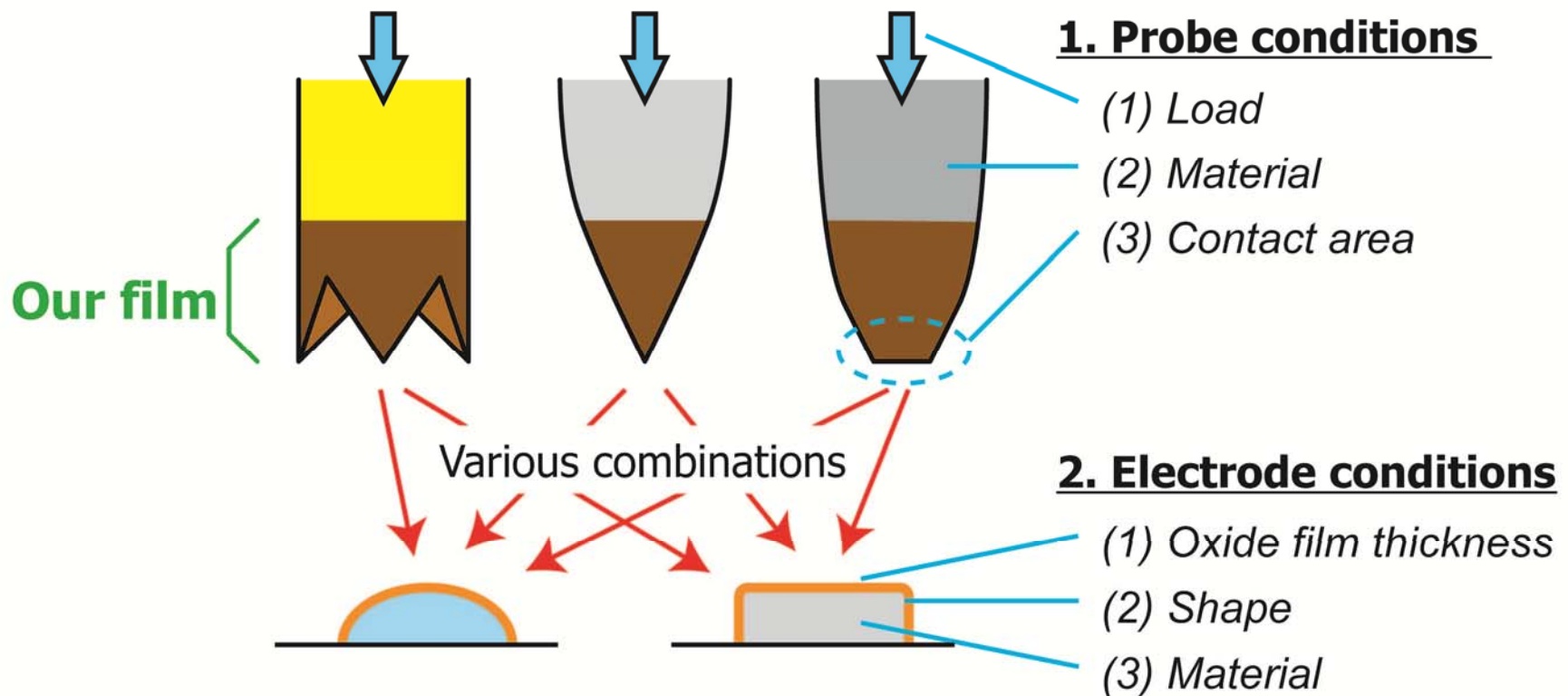
- **Why does this film have lower surface free energy?**
 - A characteristic of carbon with >99% purity
 - Higher bonding strength

Care points of practical use

4. ARE THERE ANY CARE POINTS OF PRACTICAL USE?

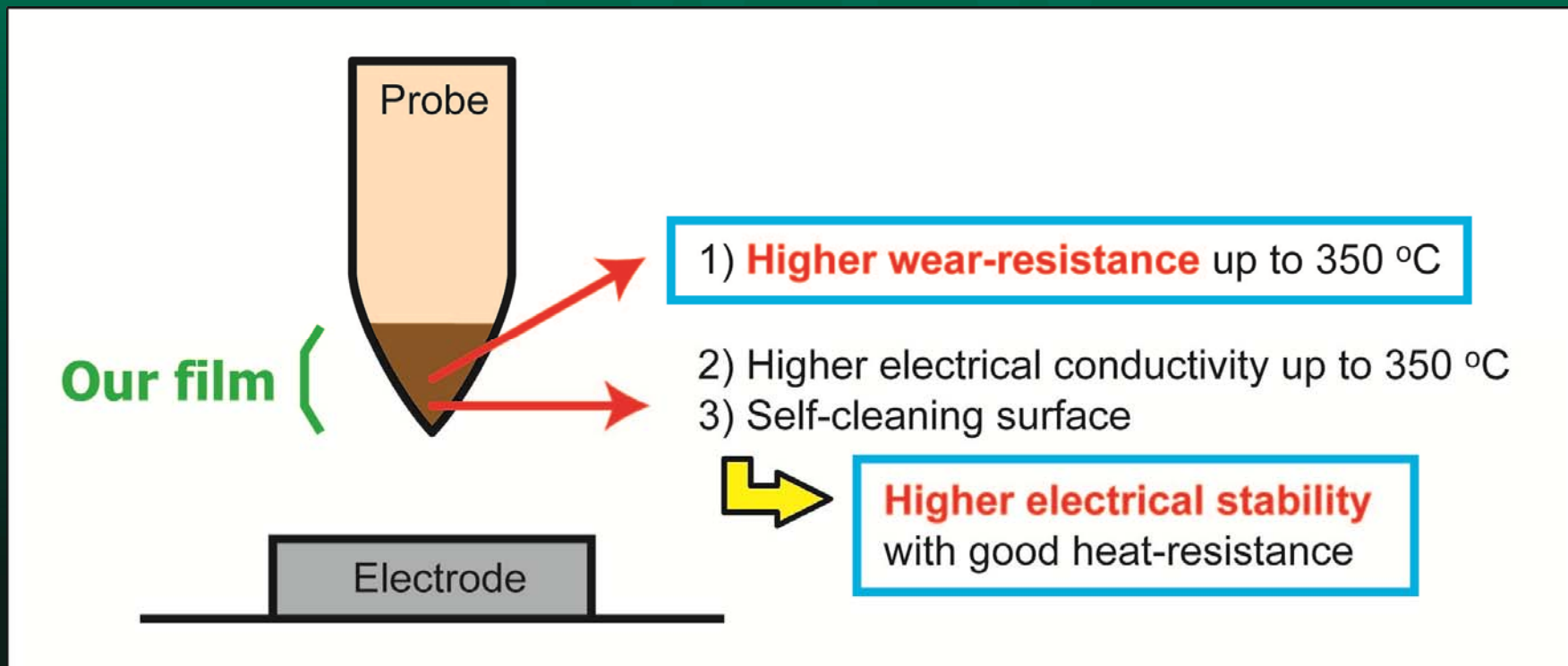
Yes, there is a care point of practical use.

- The evaluation of the film under each condition of probe user is necessary, before practical use.



5. CONCLUSION

- A practical use of the film will be further progressed, owing to the capability for longer life and better measurement reliability of probes.



- **Contact**

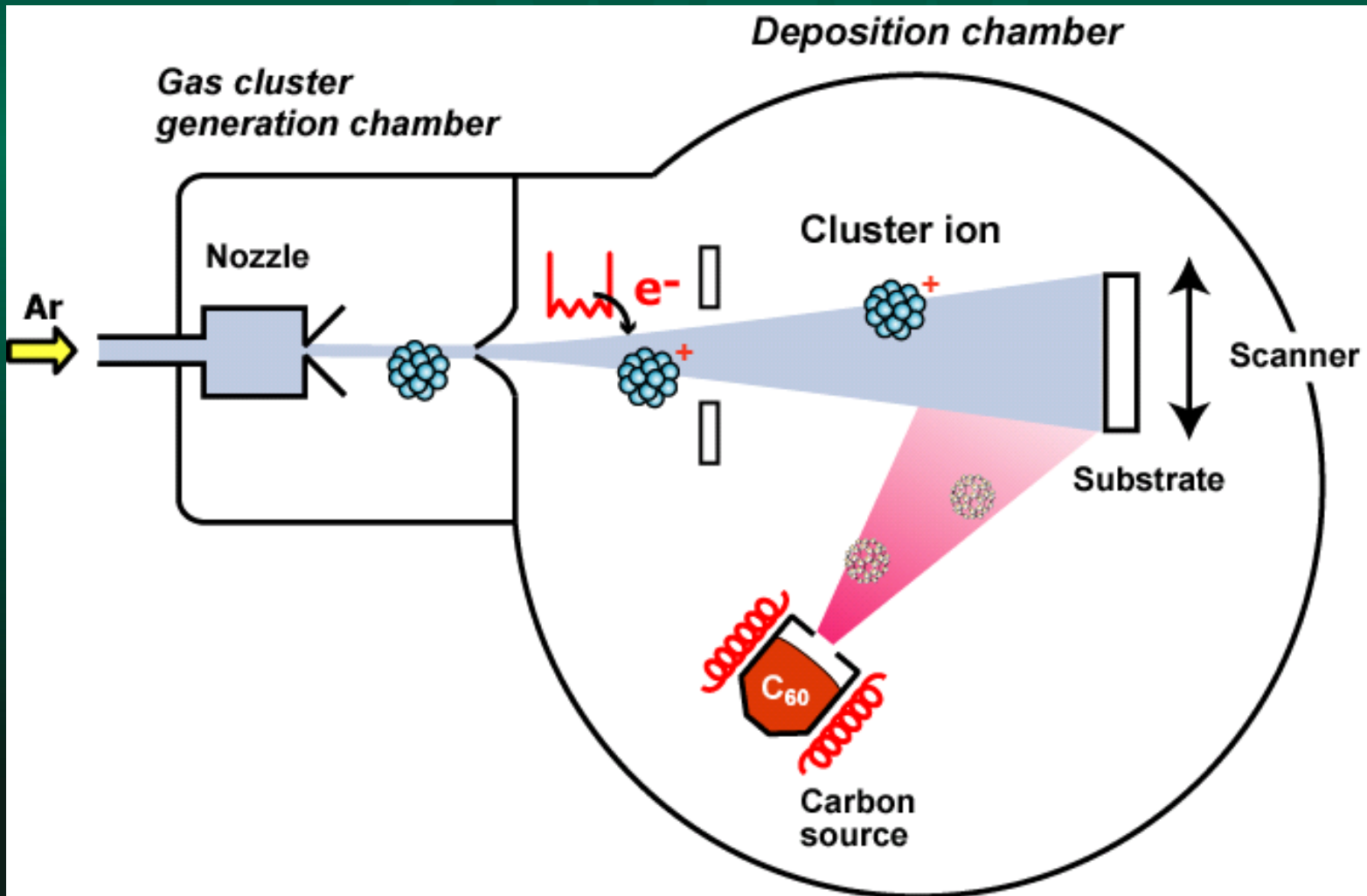
- We exhibit the SWTW Expo.
- Booth No.3

Acknowledgement

- We are deeply grateful to K. Takizawa, S. Ono, and Y. Kawahata at JC Electronics Corporation.

- for Q&A session

Gas cluster ion beam process.



High density bombardment effect of cluster ion is effective in depositing higher hardness film.

